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SYNTHESIS AND ANTIOXIDANT ACTIVITY OF N-(4-ACETAMIDOPHENYL)-2-SUBSTITUTED PHENOXY ACETAMIDE AND ACETATE DERIVATIVES; EFFECTS ON PROTEINASE ENZYME TRYPSIN AND ALBUMIN DENATURATION

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ABSTRACT

Antioxidant ability of a molecule offers protective effects against life threatening disease such as cancer, neurodegenerative disorders and obesity. The antioxidant property when combined with antiniflammatory effects, proved beneficial effects in Alzheimer's and other diseases. Considering these findings, a novel series of N-(4-acetamidophenyl)-2-substituted phenoxy acetamide and acetate derivatives were synthesized and tested for antioxidant and in vitro antiinflammatory activities. Paracetamol exhibits promising antioxidant and antiniflammatory activities. Paracetamol or its bioisostere 4-aminoacetanilide and the phenoxy (substituted with 3,4-(CH₃)₂) acid and phenoxy propionic acid were reacted to obtain a series of amide and ester derivatives. Compound IVa, 4acetamidophenyl 2-phenoxyacetate displayed marked free radical scavenging activity in DPPH assay with IC₅₀ value of 11.97±0.48µM, comparable to the standard ascorbic acid (IC₅₀, $12.50\pm0.5\mu$ M). Out of the six synthesized compounds, IVa exhibited significant proteinase i.e. trypsin inhibition activity and albumin denaturant activity.

INTRODUCTION

Antioxidants are gaining significant importance in curing diseases associated with oxidative-stress such as atherosclerosis, cancer, skin aging, arthritis and neurodegenerative disorders [1-2]. Antioxidants such as vitamin A& E, docosahexaenoic acid (DHA) have shown promising results in neurodegenerative disorders including Alzheimer's Parkinson's diseases [3-4]. In the presence of free radicals cell components, lipids, proteins, and DNA undergo oxidation to release peroxides leading to oxidative stress. Oxidative stress may lead to structural and functional abnormalities in proteins. Oxidative stress and

protein aggregation combined leads to cell death causing neurodegeneration [5].In this context, NSAIDs like ibuprofen, mefenamic acid and diclofenac were tested for their efficacy in Alzheimer's diseases. Ibuprofen and diclofenac showed promising results and the mechanism of action involves protective effects neuroinflammation against probably by properties exhibiting anti-amyloid [6]. Neuroinflammation involves complex mechanisms involving release of free radicals, cytokines (IL-1β, IL-18 & IL-33) chemokines (CCL2 &CCL5) and the activation of microglia [7-10]. These findings evidences that an agent with combined properties of antioxidant and antiinflammatoy agent might be beneficial in neurodegenerative disorders. Phenoxy derivatives are bioactive compounds several interesting activities viz., antiantioxidant, inflammatory, antibacterial, analgesic, antiplatelet, antimycobacterial, and diuretic activities [11-17].Prashanth et al reported potent antioxidant molecules. containing phenoxy acid moiety (1) in their structures. Phenoxy acetic or propionic acid moiety is also present as a pharmacophoric element in important therapeutic agents [18-19]. 2-Phenoxy-N-phenylacetamide scaffold is associated antimicrobial, anticancer. inflammatory activity and ability to inhibit Pgp efflux transporters. Paracetamol (2) a potent antipyretic and antinociceptive agent exhibits antioxidant activity. promising Several paracetamol derivatives found to possess antitubercular, antifungal and antibacterial activity [20-23].

OH
$$\frac{NH_2}{OH}$$
OH $\frac{NH_2}{HN}$
OH $\frac{NH_2}{OH}$
OH \frac

In view of the potentiality of phenoxy acid derivatives, it was planned to synthesize a series of phenoxy ester derivatives using paracetamol. Considering 4-amino acetanilide (3) as boisosteric replacement for paracetamol, a series of phenoxy amide derivatives were synthesized.

MATERIALS AND METHOD:

Phenoxy acetic acid, phenoxy propionic acid and curcumin were purchased from Himedia and paracetamol from SD fine chemicals Mumbai. Sigma melting point apparatus and **Systronic UV-VIS** Spectrophotometer-117 were used to determine melting points and absorbance. A mixture of methanol and chloroform was used as mobile phase to trace the completion of the reaction.IR spectra of the compounds were obtained using Bruker FTIR spectrophotometer (cm⁻¹)and ¹H NMR was recorded on Bruker-400.130 MHz and Jeol-400MHz.Mass (m/z) spectra were obtained using Apex Mass spectrum (300800.D).

General method for the synthesis of N-(4-acetamidophenyl)-2- substituted phenoxy acetamide derivatives (IIIa-IIIc)

Step 1: Synthesis of 3,4-dimethyl substituted phenoxy acetic acid: Chloroacetic acid (6g) was weighed accurately and dissolved in 10 ml of water to get a clear solution. 3,4-dimethylphenol (5g) was dissolved in 10 ml of 25% w/vsodium hydroxide separately in round bottomed flask. The acid solution was slowly added to the phenolic solution and the obtained mixture was heated for 6 hr at 75-80°C. Upon cooling, a solid product was obtained. After drying, it was used in the next step without further purification [18].

Substituted Phenol

Substituted phenoxy acid

R=3,4-dimethyl

Step 2: Synthesis of various phenoxy acid chlorides: Various phenoxy acid chlorides were obtained by refluxing equimolar amounts of phenoxy acetic acid/3,4-dimethoxy phenoxy acetic acid/2-phenoxy propionic acid (0.01 mol) and thionyl chloride (0.01mol) for around two hours.

Substituted phenoxy acid
$$R_1 = H$$
, $R_2 = H$, $R_3 = H$, $R_4 = H$, $R_4 = H$, $R_4 = H$, $R_5 = H$, $R_6 =$

Step 3: Synthesis of N-(4-acetamidophenyl)-2-substituted phenoxy acetamide derivatives (IIIa-IIIc):

Green synthetic approach, a neat phase trituration was employed to obtain phenoxy acetamide derivatives. Equimolar amounts4-amino acetanilide (0.01 mol) and phenoxy acid chloride/3, 4-dimethoxy phenoxy acetic acid chloride/2-phenoxy propionic acid chloride (0.01 mol) were triturated in a mortar for at least 1 hr. Completion of the reaction was monitored by TLC and then ice was added.

Then the reaction mixture was filtered and General method for the synthesis of 4acetamido phenyl substituted-2phenoxyacetate (IVa-IVc): Ethylchloroformate was added drop wise to a cooled solution (0-5°C) of phenoxy acetic acid triethylamine (1.02g,0.01mol) dichloromethane (50ml). The mixture was stirred for 30 min and then paracetamol (1.51g, 0.01mol) was added over a period of] % inhibition (Absorbance of control-Absorbance of test) / Absorbance of control× 100

R= H; 3,4-dimethyl R₁= H, CH₃

Nitric oxide scavenging assay: The test samples (5, 10, 20, 40, 80μm) were prepared using methanol. 2 ml Sodium nitroprusside (10μm) in phosphate buffer pH 7.4 was incubated with 2ml of test sample at 25°C for 120 min. Curcumin was considered as positive control. The absorbance of test samples and standard drug was read at 546nm using methanol as blank [27]

2, 2'- Azino- bis (3-ethylbenzothiazoline-6sulfonic acid) diammonium salt (ABTS) reduction assay

Test samples of $100\mu m$ concentration were prepared using ethanol. To 1ml of test sample 1ml of ABTS solution was added and mixed thoroughly. After 15 min, the absorbance values were measured at 734 nm. Similarly 5, 10, 20, 40, 80 μM concentrated test solutions were prepared and the absorbance values were noted to determine IC₅₀ values [28].

IN VITRO ANTIINFLAMMATORY STUDIES

Protein Inhibitory Action: Proteinases play an important role in the tissue damage during inflammation. 0.06 mg Trypsin and 1 ml of 20mM Tris-HCl buffer (pH 7.4) were added with ml test sample of different 1 concentrations $2\mu M$, $4\mu M$, $(1\mu M,$ 16µM). The reaction mixture was incubated at 37°C for 5 min and then 1ml of 0.8% (W/V) dried [24].

casein was added. The mixture was incubated for an additional 20 min then 2 ml of 70% perchloric acid was added to terminate the reaction. A suspension was obtained which was centrifuged and the absorbance of the supernatant was read at 210 nm against buffer as blank. The experiment was performed in triplicate. The percentage of inhibition of proteinase inhibitory activity was calculated [29]. % inhibition = (Absorbance of control-Absorbance of test) / Absorbance of control× 100

Protein Denaturation Activity

Denaturation of protein is considered as one of the reasons of inflammation. The reaction mixture (5 mL) consisted of 0.2 mL of egg albumin (hen's egg), 2.8 mL of phosphate buffered saline (PBS, pH 6.4) and 2 mL of varying concentrations of test samples (1 - 16µm). Distilled water served as control. The mixtures were incubated at 37°C for 15 min and then heated at 70°C for 5 min. After cooling, absorbance was measured at 660 nm using methanol as blank. Aspirin was used as reference drug. The percentage inhibition of protein denaturation was calculated [30].

% inhibition = (Absorbance of control-Absorbance of test) / Absorbance of control× 100

RESULTS AND DISCUSSION Chemistry

N-(4-acetamidophenyl)-2-substituted phenoxy acetamide derivatives (IIIa-IIIc) were synthesized by treating various substituted phenoxy acid chlorides with 4aminoacetanilide. Different phenoxy acids were coupled with paracetamol usingethylchloroformate andtriethylamine to N-(4-acetamidophenyl)-2-substituted phenoxy acetate derivatives. Phenoxy acetic acid / 3, 4-dimethylphenoxy acetic acidwas reacted with chloro acetic acid to give substituted phenoxy acids while 2-phenoxy propionic acid was purchased from Sigma Chemicals, India. For the conversion of acids to acid chlorides, thionyl chloride was used. After synthesizing the acid chlorides, they were triturated in solvent free conditions with 4amino acetanilide to form IIIa-IIIc. The amide analogues were obtained in good yields (63-81%) when compared to ester analogues (3375%). In the IR spectra, absorption bands due to N-H stretch and C=Ostretch were observed around 3300 cm $^{-1}$ and 1660 -1700 cm $^{-1}$ in amides, whereas in ester derivatives, C=O and C-O-C stretch around 1700 cm $^{-1}$, 1010-1096 cm $^{-1}$ respectively. In the NMR spectra, aromatic protons were observed at around 7-8ppm and phenoxy methylene protons were observed at around 4.8 to 5.0 ppm. The titled compounds showed characteristic M $^+$ or [M+1] peaks.

In vitro antioxidant studies

Among all the compounds, compound IVa displayed marked radical scavenging activity with IC₅₀value of 11.97µM in DPPH assay while ascorbic acid displayed high ability (IC_{50.} 12.50 µM)to inhibit the reduction of DPPH radical. Activity profile of the title compounds in nitric oxide scavenging assay is similar to that of DPPH assay. Compound III c, which displayed moderate activity in DPPH assay, also displayed moderate activity in NO scavenging assay. A perusal of data in ABTS assay indicates that IIIa, IIIc and IVa exhibited less activity when compared to the standard compound BHT. Antioxidant profile of title compounds highlighted IVa which possess paracetamol moiety in its structure. Paracetamol has strong antioxidant properties due to its ability to transfer hydrogen or electron to DPPH free radical. Free radical scavenging ability of paracetamol was reported to be increased with the introduction of indazole or N-methylmophonium group on acyl moiety [21].

4-Acetamidophenyl 2-phenoxyacetate

RESULTS OF ANTI-INFLAMMATORY STUDIES

Proteinase inhibition

In this method, trypsin inhibits leucocytes proteinase which plays an important role in the development of tissue damage during antiinflammatory activity. Compounds IVa, IVc and IIIb showed promising inhibition in this method.

Protein denaturation

NSAIDs have ability to protect albumin proteins against denaturation. like antidenaturant effect plays important role in chronic inflammatory conditions such as rheumatoid arthritis and neurodegenerative diseases. Aspirin and majority of the NSAIDs (Ibuprofen, mefenamic acid) reported to exhibit neuroprotective activity in conditions like Alzheimer's disease possibly due to their protective action on proteins [31]. These experimental evidences suggested this model as a reliable screening model for new antiinflammatory drugs [32-34]. Compound IVb exhibited good protecting ability compared to other compounds.

Table. 1 *In vitro* antioxidant studies (IIIa – IIIc & IVa-IVc)

Compound No	IC ₅₀ (DPPH scavenging	IC ₅₀ (NO scavenging	IC ₅₀ (ABTS assay) μM
	assay) μM	assay)	
		${f \mu M}$	
IIIa	43.61±2.12	63.08±2.01	66.21±2.42
IIIb	67.32±2.28	-	-
IIIc	22.70±1.98	26.45±1.02	34.25±1.54
IVa	11.97±2.24	41.70±2.04	58.95±1.62
IVb	67.96±3.01	-	-
IVc	54.61±1.86	-	-
Standard	Ascorbic acid	Curcumin	Butylated Hydroxy
	12.5±0.42	16.54 ± 0.88	Toluene (BHT)
			21.32±0.62

Table.2 In vitro anti-inflammatory studies (IIIa - IIIc& IVa-IVc)

Compound No	IC ₅₀ (proteinase inhibition) μM	IC ₅₀ (protein denaturation) μM
IIIa	45.25±1.86	36.32±1.62
IIIb	36.25±1.20	41.03±0.94
IIIc	42.32±2.04	34.52±2.02
IVa	32.28±1.48	37.15±1.55
IVb	41.28±1.62	31.25±1.24
IVc	37.14±1.88	46.24±1.86
Standard	Aspirin (5µM)	Ibuprofen (5μM)
(Positive control)	63.14±2.68	12.14±1.80

CONCLUSION

The synthesized compounds showed antioxidant and in promising vitro antiinflammatory activity. Among all 4-acetamidophenyl Compound IVa, 2phenoxyacetate displayed marked free radical scavenging activity in DPPH assay with IC₅₀ value of 11.97±0.48µM, comparable to the standard ascorbic acid (IC₅₀, 12.50±0.5µM). The title compounds also displayed remarkable in vitro antiniflammatory activity suggesting their potential in treating neurodegenerative disorders.

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